Urganoiodine(III)-Mediated Efficient Synthesis of New 3,9-DiaryI-bis-1,2,4-triazolo[4,3*a*][4,3-*c*]pyrimidines

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ABSTRACT: Oxidation of bis-2,4-pyrimidinylhydrazones of various araldehydes with two equivalents of iodobenzene diacetate in dichloromethane provides an efficient and easy method for the synthesis of eight new 3,9-diaryl-bis-1,2,4-triazolo[4,3-a][4,3c]pyrimidines. © 2006 Wiley Periodicals, Inc. Heteroatom Chem 17:653–655, 2006; Published online in Wiley InterScience (www.interscience.wiley.com). DOI 10.1002/hc.20250

INTRODUCTION

In recent years, organohypervalent iodine reagents [1] have emerged as reagents of choice for various synthetically useful transformations due to their low toxicity, ready availability, and ease of handling. As part of our ongoing program directed toward the unique applications of organohypervalent iodine reagents in organic synthesis, we have shown that organoiodine(III) reagents such as iodobenzene diacetate (IBD) [2] and [hydroxy(tosyloxy)iodo]benzene (HTIB, Koser's reagent) [3] are excellent reagents for the oxidation of phenolic compounds [4], α -functionalization

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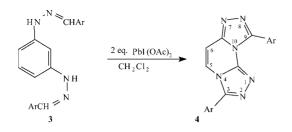


of carbonyl compounds [5], and synthesis of a wide variety of heterocyclic compounds [6], including biologically active compounds [7]. In continuation of these encouraging reports, we have developed an efficient and easy method for the synthesis of new 3,9-diaryl-bis-1,2,4-triazolo[4,3-*a*][4,3-*c*]pyrimidines **4** as potential antibacterial agents. The results of this useful synthesis involving oxidation of bis-hydrazones **3** with IBD are reported herein.

Oxidative cyclization of hydrazones of general formula **1** using various reagents is a viable route for the synthesis of fused triazolo heterocyclic systems of the type **2** [8].

$$\left\{ \underbrace{\bigwedge_{N}}_{H}^{R} \xrightarrow{} \left\{ \underbrace{\bigwedge_{N}}_{N}^{R} \xrightarrow{} \left\{ \underbrace{\bigwedge_{N}}_{N}^{R} \xrightarrow{} \right\} \right\} \right\}$$
(1)

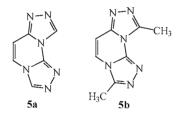
Based on synthetic route given in Eq. (1), it was anticipitated that oxidation of bis-hydrazones **3** with two equivalents of IBD might afford the desired bistriazoles **4**. Accordingly, a solution of bis-hydrazone **3a** in dichloromethane was treated with two equivalents of IBD at room temperature. A rapid reaction occurred to afford the product **4a** as crystalline solid in 60% yield. Encouraged by the feasibility of our strategy for **4a**, we studied oxidative cyclization of a wide range of substituted hydrazones **4b–h** with IBD under similar conditions. The method, indeed, worked efficiently for the synthesis of aryl derivatives **4b–h** (Scheme 1, Table 1).



SCHEME 1

The mechanism of this transformation is not certain, and a probable mechanism is outlined in Scheme 2. The first step involves the electrophilic attack of IBD on **3** to form an organoiodine(III) intermediate **A**. Subsequently, **A** generates another intermediate, a bis-nitrile amide **B** along with expulsion of two molecules of iodobenzene and acetic acid each. The nitrile amide **B** undergoes cyclization to give the product **4**.

It should be noted that synthesis of parent bis-1,2,4-triazolo[4,3-*a*][4,3-*c*]pyrimidine **5a** and its 3,9dimethyl derivative **5b** has been reported earlier [9], whereas no example of 3,9-diaryl derivative of the type **4** is known yet.

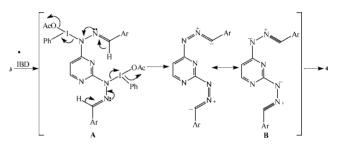


The reported synthetic route to **5a** and **5b** involves the reaction of 2,4-dihyrazinopyrimidine **6** with formic acid and triethyl orthoacetate, respectively. In an analogous manner, we attempted the preparation of **4a** by the reaction of **6** with benzoic acid/benzoyl chloride under various conditions [10]. Although we were successful in obtaining dibenzoylamino derivative **7**, the cyclization of the latter did not give any acceptable result to obtain the desired product **4a** (Scheme 3).

Finally, the present study offers an easy access to new bis-triazolopyrimidines **4**, which are potential antibacterial agents [11]. The method for oxidative cyclization of hydrazones **3** to triazolo heterocyclic compounds **4** involves mild conditions.

EXPERIMENTAL

Melting points were determined in open capillaries in electrical melting point apparatus and are



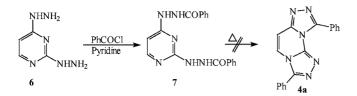
SCHEME 2

TABLE 1 3,9-Diaryl-bis-1,2,4-triazolo[4,3-a][4,3-c]pyrimidines 4 Prepared According to Scheme 1

Product ^a	Ar	М.р. (°С)	Yield ^b (%)	¹ Η NMR (CDCl ₃) δ
4a	C ₆ H ₅	82–85	60	7.81 (d, 1H, J=8.1 Hz), 7.31 (d, 1H, J=8.1 Hz), 7.2–7.55 (m, 6H), 7.95–8.10 (m, 4H)
4b	4 -CH $_3$ C $_6$ H $_4$	98–102	62	2.50 (s, 6H, CH ₃), 7.86 (d, 1H, J = 8.1 Hz), 7.25 (d, 1H, J = 8.1 Hz), 8.18 (d, 2H, J = 8.4 Hz), 7.68 (d, 2H, J = 8.4 Hz), 7.68 (d, 2H, J = 8.4 Hz), 7.40–7.45 (m, 4H)
4c	4-CH ₃ OC ₆ H ₄	121–123	57	3.83 (s, 6H, OCH_3), 7.9 (d, 1H, $J = 8.1$ Hz), 7.62 (d, 1H, $J = 8.1$ Hz), 8.17 (d, 2H, $J = 8.7$ Hz), 7.61 (d, 2H, $J = 8.7$ Hz), 7.61 (d, 2H, $J = 8.7$ Hz), 7.02–7.05 (m, 2H), 6.90–6.94 (m, 2H)
4d	2,5-(MeO) ₂ C ₆ H ₃	92–96	67	3.85 (s, 12H, OCH ₃), 7.62 (d, 1H, <i>J</i> =8.1 Hz), 7.28 (d, 1H, <i>J</i> =8.1 Hz), 7.12–7.23 (m, 6H)
4e	4-FC ₆ H ₄	192–195	59	7.90 (d, 1H, J=8.1 Hz), 7.36 (d, 1H, J=8.1 Hz), 8.38–8.45 (m, 2H), 8.24–8.29 (m, 2H), 7.18–23 (m, 4H)
4f	4-CIC ₆ H ₄	202–206	67	7.85 (d, 1H, $J = 8.1$ Hz), 7.62 (d, 1H, $J = 8.1$ Hz), 8.35 (d, 2H, $J = 8.7$ Hz), 8.20 (d, 2H, $J = 8.7$ Hz), 7.51–7.56 (m, 4H)
4g	4-BrC ₆ H ₄	150–153	62	7.94 (d, 1H, J=8.1 Hz), 7.64 (d, 1H, J=8.1 Hz), 8.30 (d, 2H, J=8.7 Hz), 8.14 (d, 2H, J=8.7 Hz), 7.68–7.75 (m, 4H)
4h	$4-O_2NC_6H_4$	183–185	68	8.16 (d, 1H, $J = 8.1$ Hz), 8.02 (d, 1H, $J = 8.1$ Hz), 8.4–8.5 (m, 4H), 8.11–8.25 (m, 4H)

^aThe elemental analyses (C, H, N) of all the products were found within the $\pm 0.4\%$ of their calculated values.

^bThe yields of the isolated pure products 4 with respect to 3 under the optimized conditions.



SCHEME 3

uncorrected. The IR (KBr) and ¹H NMR spectra were recorded on Buck Scientific IR M-500 and Bruker (300 MHz) spectrophotometers, respectively. Bis-2,4-pyrimidinylhydrazones **3** were synthesized by refluxing 2,4-dihydrazinopyrimidine **6** and araldehydes in ethanol with a trace of glacial acetic acid on water bath for about 2–3 h [12,13].

3,9-Diaryl-bis-1,2,4-triazolo[4,3-a][4,3c]pyrimidines **4**

General Procedure. To a stirred solution of hydrazones **3** (0.01 mol) in dichloromethane (25 mL) at room temperature, IBD (0.02 mol) was added in four to five portions during 5 min. The solvent was evaporated on a steam bath and the residual mass containing product and iodobenzene triturated with petroleum ether to give solid product, which was recrystallized from methanol to yield pure bistriazolopyrimidines **4**.

REFERENCES

 For selected reviews see: (a) Moriarty, R. M. J Org Chem 2005, 70, 2893–2904; (b) Koser, G. F. Adv Heterocycl Chem 2004, 86, 225–292; (c) Wirth, T. Hypervalent iodine chemistry: Modern developments in organic synthesis; In Topics in Current Chemistry; Wirth, T. (Ed.); Springer: Berlin, 2003; Vol. 224; (d) Zhdankin, V. V.; Stang, P. J. Chem Rev 2002, 102, 2523–2584; (e) Ochiai, M. In Chemistry of Hypervalent Compounds; Akibia, K. (Ed.); VCH: New York, 1999; Ch. 13, pp. 359–387; (f) Varvoglis, A. Hypervalent Iodine in Organic Synthesis; Academic: New York, 1997; (g) Kita, Y.; Tohma, H.; Yakura, T. Trends Org Chem 1992, 3, 113.

- [2] (a) Moriarty, R. M.; Chany, C. J., II; Kosmeder, J. W., II. In Encyclopedia of Reagents in Organic Synthesis; Paquette, L. A. (Ed.); John Wiley & Sons: Chichester, 1995; Vol. 2, pp. 1479–1484; (b) Prakash, O.; Singh, S. P. Aldrichim Acta 1994, 27, 15–22; (c) Varvoglis, A. Chem Soc Rev 1981, 10, 377–407.
- [3] (a) Koser, G. F. Aldrichim Acta 2001, 34, 89–101;
 (b) Prakash, O.; Saini, N.; Sharma, P. K. Heterocycles 1994, 38, 409–430;
 (c) Moriarty, R. M.; Vaid, R. K.; Koser, G. F. Synlett 1990, 365–383.
- [4] (a) Prakash, O.; Sharma, V.; Bhardwaj, V.; Pundeer, R.; Kaur, H. Tetrahedron Lett 2004, 45, 9065–9067; (b) Prakash, O.; Pundeer, R.; Kaur, H. Synthesis 2003, 2768–2770; (c) Prakash, O. Sharma, V.; Tanwar, M. P. Can J Chem 1999, 77, 1191–1195.
- [5] Prakash, O.; Kaur, H.; Batra, H.; Rani, N.; Singh, S. P. J Org Chem 2001, 66, 2019–2023.
- [6] (a) Prakash, O.; Batra, A.; Sharma, V.; Saini, R. K.; Varma, R. S. J Ind Chem Soc 2003, 80, 1031–1033; (b) Prakash, O.; Sharma, P. K.; Saini, N. Synlett 1994, 221–227.
- [7] (a) Sadana, A.; Mirza, Y.; Aneja, K. R.; Prakash, O. Eur J Med Chem 2003, 38, 533–536; (b) Prakash, O.; Bhardwaj, V.; Kumar, R.; Tyagi, P.; Aneja, K. R. Eur J Med Chem 2004, 39, 1073–1077.
- [8] (a) Kumar, D.; Prakash, O.; Singh, S. P. J Chem Res (S) 1993, 244–245; (b) Gladstone, W. A. F.; Aylward, J. B.; Norman, R. O. C. J Chem Soc C 1969, 2587; (c) Gibson, M. S. Tetrahedron 1963, 19, 1587–1589.
- [9] (a) Brown, D. J.; Nagamatsu, T. Aust J Chem 1979, 32, 1585–1593; (b) La Noce, T.; Guliani, A. M. J Heterocycl Chem 1975, 12, 551–555.
- [10] Pollack, A.; Tisler, M. Tetrahedron 1966, 22, 2073–2079.
- [11] Preliminary results on the antibacterial screening have shown that some of these compounds are associated with strong activity.
- [12] Fabbrini, L. Guzz Chim Ital 1957, 87, 1293–1302 (University of Florence).
- [13] Vanderhaeghe, H.; Claesen, M. Bull Soc Chim Belg 1959, 68, 30–46 (University of Louvain).